

Stoichiometry Problem Sheet 1 Answers File Type

Stoichiometry: Problem Sheet 1

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Answers To Stoichiometry Worksheet

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Answers: Moles and Stoichiometry Practice Problems 1) How many moles of sodium atoms correspond to 1.56×10^{21} atoms of sodium? 1.56×10^{21} atoms Na $\times 1 \text{ mol Na} = 2.59 \times 10^{-3} \text{ mol Na}$ 236.022×10 atoms Na 2) Determine the mass in grams of each of the following: a. 1.35 mol of Fe 1.35 mol Fe $\times 55.845 \text{ g Fe} = 75.4 \text{ g Fe}$ 1 mol Fe b. 24.5 mol O

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Solution Stoichiometry Worksheet. Solve the following solutions Stoichiometry problems: 1. How many grams of silver chromate will precipitate when 150. mL of 0.500 M silver nitrate are added . to 100. mL of 0. 400 M potassium chromate? $2\text{AgNO}_3(aq) + \text{K}_2\text{CrO}_4(aq) \rightarrow \text{Ag}_2\text{CrO}_4(s) + 2\text{KNO}_3(aq)$ 2.

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Stoichiometry Worksheet 1 Mole To Mole Calculations Answers

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