

# Section 53 Electron Configurations

## Answers

38. The ground state electron configuration of Fe is \_\_\_\_\_. A.  $2 1s 2s 2 3s 3p^6 3d^6$  B.  $2 1s 2 2s 2p^6 3s^2 3p^6 4s 3d^6$  C.  $2 1s^2 2s 2p^6 3s^2 3p^6 4s^2$  D.  $2 1s 2 2s 2p^6 3s 3p^6 4s^2 4d^6$
39. The ground state electron configuration of Ga is \_\_\_\_\_. A.  $2 1s 2s^2 3s^2 3p^6 3d^{10} 4s^2 4p^1$  B.  $6 1s 2 2s 2p 3s^2 3p^6 4s^2 4d^{10} 4p^1$  C.  $2 1s^2 2s 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^1$

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Actual Electron Configurations. • Total electrons = atomic  
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number • Fill energy levels with electrons until you run out

- A superscript states how many electrons are in each level.
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### Electron Configuration (Section 5.2)

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Answer: All have an  $ns^2 np^5$  electron configuration, one electron short of a noble gas electron configuration. (Note that the heavier halogens also have filled  $(n - 1)d^{10}$  subshells, as well as an  $(n - 2)f^{14}$  subshell for Rn; these do not, however, affect their chemistry in any significant way.

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permit/deny {service/protocol} {source network/IP}

{destination network/IP} {port#} Named ACL: Ip access-list

extended NAME Permit x.x.x.x x.x.x.x Deny x.x.x.x x.x.x.x

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The electron configuration of an atom is  $1s^2 2s^2 2p^6$ . The number of valence electrons in the atom is The number of valence electrons in the atom is answer choices

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## Atomic Structure and Electron Configurations

### Multiple ...

Section 53 Study Guide Answers Section 53

Answers. Standard ACL: access-list x permit host y.y.y.y or access-list x permit x.x.x.x x.x.x.x

Extended ACL: access-list x permit/deny  
{service/protocol} {source network/IP} {destination  
network/IP} {port#} Named ACL: Ip access-list  
extended NAME Permit x.x.x.x x.x.x.x Deny x.x.x.x  
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Answers

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answerElectron configuration questionElectrons in



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