

Modern Chemistry Chapter 7 Test B Answers

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7 Chemical Formulas and Chemical Compounds

CHAPTER 7 REVIEW. Chemical Formulas and Chemical Compounds. SECTION 1. SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us. (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. Modern Chemistry 90 Chapter Test Name Class Date Chapter Test B, continued PART III Match the letter on the diagram with the term for that point or region. 18. Critical point 19. Vapor 20. Normal freezing point 21. Triple point 22. Solid 23. Normal boiling point 24. Liquid 25. Critical pressure PART IV Write the answers to the following questions in the space provided. 26.

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Modern Chemistry Chapter 7 test. STUDY. PLAY. The basis of an ionic bond is the___ absorption of water into their solid structures. Elements tend to gain or lose electrons in order to acquire the electron configuration of a___ noble gas. Ions that are formed from more than one atom are called___

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Modern Chemistry Chapter 7. compounds composed of two different elements; (compounds of 2 elements between metal (positive, s-block and p-block) and nonmetal (negative); metal is always listed first; change nonmetal to -ide ending)

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Rules 1-7 apply covalently bonded atoms; Oxidation numbers can also be assigned to atoms in ionic compounds. Oxidation Rule 9 Monoatomic ion oxidation number equal to ion charge Na+, Ca2+, Cl-

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