

# **Modern Chemistry**

## **Answer Key Chapter 10**

Modern Chemistry 105 Chapter  
Test Name Class Date Chapter  
Test A, continued Use this  
figure to answer questions 7

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and 8. \_\_\_\_\_ 7. A solution containing 35 g of  $\text{Li}_2\text{SO}_4$  dissolved in 100 g of water is heated from  $10^\circ\text{C}$  to  $90^\circ\text{C}$ . According to information in the figure, this temperature change would result in a. an additional 5

g of Li<sub>2</sub>SO<sub>4</sub> in ...

Modern Chemistry Chapter 6 4  
Answers

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chemistry answer key chapter  
2. a 2. b 3. c 4. c 5. a 6. b 7  
Modern chemistry answer key  
chapter 2. c 8. Polarity in a  
water molecule is caused by an

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uneven distribution of electrons between the oxygen and hydrogen atoms. 9. The concentration of H ions determines whether a solution is acidic or basic.

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ELECTRONS IN ATOMS The  
following pages contain the bulk  
(but not all) of the information

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CHAPTER 2 REVIEW  
Measurements and Calculations

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## SECTION 1 SHORT ANSWER

Answer the following questions in the space provided. 1.

Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a

model. observation and data a.  
A research team records the  
rainfall in inches

## 2 Measurements and Calculations

### Modern Chemistry 105 Chapter

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Test Name Class Date Chapter  
Test A, continued Use this  
figure to answer questions 7  
and 8. \_\_\_\_\_ 7. A solution  
containing 35 g of  $\text{Li}_2\text{SO}_4$   
dissolved in 100 g of water is  
heated from  $10^\circ\text{C}$  to  $90^\circ\text{C}$ .

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According to information in the figure, this temperature change would result in a. an additional 5 g of  $\text{Li}_2\text{SO}_4$  in ...

Assessment Chapter Test A -  
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2O shows that 2 molecules (made of 4 atoms) of hydrogen and 1 molecule (made of 2 atoms) of oxygen produce 2 molecules of water. The total mass of the product, water, is equal to the sum of the masses

of each of the reactants, hydrogen and oxygen. What parts of Dalton ' s atomic theory are illustrated by this reaction?

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to the questions on the line to  
the left, and show your work in  
the space provided. 25. 26.

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atomic number of nickel-60 is 28. How many neutrons does this iso- tope have? Carbon-14 has 8 neutrons. What is the atomic number of carbon-14? 11

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CHAPTER 1 REVIEW Matter  
and Change MIXED REVIEW  
SHORT ANSWER Answer the

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following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance.

homogeneous a. sugar

homogeneous d. plastic wrap

homogeneous b. iron filings

heterogeneous e. cement  
sidewalk heterogeneous c.  
granola bar 2. For each type of  
investigation, select the most  
appropriate ...

## 1 Matter and Change -

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of Matter SECTION 3 SHORT

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ANSWER Answer the following questions in the space provided.

1. Match description on the right to the correct crystal type on the left.

b ionic crystal (a) has mobile electrons in the crystal c covalent molecular

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crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

10 States of Matter - Ms.  
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CHAPTER 22 REVIEW Organic  
Chemistry SECTION 1 SHORT  
ANSWER Answer the following

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questions in the space provided.

1. Name two types of carbon-containing molecules that are not organic. \_\_\_\_\_ 2. \_\_\_\_\_

Carbon atoms form bonds readily with atoms of ...

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and Calculations SECTION 1 SHORT  
ANSWER Answer the following questions

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in the space provided. 1. Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a model.

observation and data a. A research team records the rainfall in inches

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content, but make ...

2O shows that 2 molecules (made of 4 atoms) of hydrogen and 1 molecule (made of 2 atoms) of oxygen produce 2 molecules of water. The total mass of the product, water, is equal to the sum of the

masses of each of the reactants, hydrogen and oxygen. What parts of Dalton ' s atomic theory are illustrated by this reaction?

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right to the correct crystal type on the  
left. b ionic crystal (a) has mobile

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electrons in the crystal c covalent  
molecular crystal (b) is hard, brittle, and  
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Change MIXED REVIEW SHORT  
ANSWER Answer the following

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questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance.

homogeneous	a.							
sugar	homogeneous	d.	plastic wrap	homogeneous	b.	iron filings	heterogeneous	e.

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cement sidewalk  
heterogeneous c. granola bar  
2. For each type of  
investigation, select the  
most appropriate ...

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b 7 Modern chemistry answer  
key chapter 2. c 8. Polarity  
in a water molecule is

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caused by an uneven distribution of electrons between the oxygen and hydrogen atoms. 9. The concentration of H ions determines whether a solution is acidic or basic.

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the chapter 4 test Focus on  
this content, but make ...

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Measurements and

Calculations SECTION 1 SHORT

ANSWER Answer the following  
questions in the space

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provided. 1. Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a model.

observation and data a. A research team records the

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rainfall in inches

## **2 Measurements and Calculations**

Modern Chemistry 105 Chapter  
Test Name Class Date Chapter  
Test A, continued Use this  
figure to answer questions 7

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in ...

**Assessment Chapter Test A -  
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20 shows that 2 molecules  
(made of 4 atoms) of  
hydrogen and 1 molecule  
(made of 2 atoms) of oxygen

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produce 2 molecules of water. The total mass of the product, water, is equal to the sum of the masses of each of the reactants, hydrogen and oxygen. What parts of Dalton's atomic theory are illustrated by

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this reaction?

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to the left, and show your work in the space provided.

25. 26. MODERN CHEMISTRY The atomic number of nickel-60 is 28. How many neutrons does this isotope have? Carbon-14 has 8 neutrons. What is the atomic number of

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carbon-14? 11 CHAPTER 3 TEST

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Change MIXED REVIEW SHORT

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ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. homogeneous a. sugar homogeneous d. plastic wrap homogeneous b. iron

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filings heterogeneous e.  
cement sidewalk  
heterogeneous c. granola bar  
2. For each type of  
investigation, select the  
most appropriate ...

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CHAPTER 10 REVIEW States of  
Matter SECTION 3 SHORT  
ANSWER Answer the following  
questions in the space  
provided. 1. Match

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description on the right to  
the correct crystal type on  
the left. b ionic crystal  
(a) has mobile electrons in  
the crystal c covalent  
molecular crystal (b) is  
hard, brittle, and  
nonconducting a metallic

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crystal (c) typically has the lowest melting point of the four

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Algebra 1: Common Core (15th Edition) Charles, Randall I.

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179 Organic Chemistry  
CHAPTER 22 REVIEW Organic  
Chemistry SECTION 1 SHORT  
ANSWER Answer the following  
questions in the space  
provided. 1. Name two types  
of carbon-containing  
molecules that are not

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organic. \_\_\_\_\_ 2. \_\_\_\_\_  
Carbon atoms form bonds  
readily with atoms of ...

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Ed W. Clark High School**

PROBLEMS Write the answers  
to the questions on the line  
to the left, and show your

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work in the space provided.

25. 26. MODERN CHEMISTRY The atomic number of nickel-60 is 28. How many neutrons does this isotope have? Carbon-14 has 8 neutrons. What is the atomic number of carbon-14? 11 CHAPTER 3 TEST

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ANSWER KEY 1 Modern  
chemistry answer key  
chapter 2. a 2. b 3. c 4.  
c 5. a 6. b 7 Modern  
chemistry answer key  
chapter 2. c 8. Polarity  
in a water molecule is

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caused by an uneven distribution of electrons between the oxygen and hydrogen atoms. 9. The concentration of H ions determines whether a solution is acidic or

basic.

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