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of matter 1. a) i) Ammonium chloride (1) ii) Diffusion  
Diffusion is the movement of particles from an area of

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higher concentration to an area of lower concentration until the particles are evenly distributed. (2) iii) The ammonia solution gave off ammonia gas and

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Answers in Appendix E 1. Convert a pressure of 1.75 atm to kPa and to mm Hg. 2. The critical pressure of carbon dioxide is 72.7 atm. What is this value in units of pascals? MAIN IDEA The total pressure of a gas mixture is the sum of the pressures of the gases in it. John Dalton, the English chemist who proposed the atomic theory, also

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Answers in Appendix E 1. Convert a pre sure of 1.75 atm

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to kPa and to mm Hg. 2. The critical pressure of carbon dioxide is 72.7 atm. What is this value in units of pascals? MAIN IDEA The total pressure of a gas mixture is the sum of the pressures of the gases in it. John Dalton, the English chemist who proposed the atomic theory, also

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of particles from an area of higher  
concentration to an area of lower  
concentration until the particles are evenly  
distributed. (2) iii) The ammonia solution  
gave off ammonia gas and  
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